

115. X-Ray Structure Analyses of Alkyl-Substituted *N*-Acryloyl- and *N*-Crotonoyltoluenesultams¹⁾

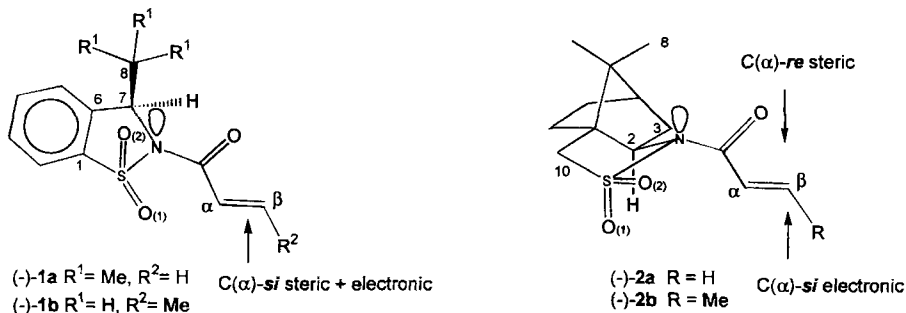
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Dedicated to the memory of Professor *Wolfgang Oppolzer*

(14. IV. 97)

The X-ray crystal-structure analyses of two *N*-enoyl-toluenesultam derivatives **1** are reported. The structural differences with their bornanesultam analogues **2** as well as stereochemical aspects with respect to their [4 + 2] and [3 + 2] cycloadditions are discussed.

The preparation as well as the [4 + 2] and [3 + 2] cycloadditions of *N*-enoyltoluenesultams (–)-**1a,b** have been recently reported [1] [2]³⁾, simultaneously with the α -alkylation and aldolization of *N*-acyl derivatives [4]. The X-ray structure analyses of the camphor analogues (–)-**2a,b** [5] [6] have been extensively used to rationalize their reactive conformations [7–9]. Both bornane [6] and toluenesultam auxiliaries [10] have been speculatively compared to a disguised C_2 -symmetrical 2,5-dimethylpyrrolidine system [8], where the approach on the enoyl side chain is sterically disfavoured either by the C-skeleton in the $SO_2/C(O)$ -*syn*, $C(O)/C=C$ -*s-cis* conformation (*syn-s-cis*) or by the O(1) atom in the *anti-s-cis* arrangement [6] [8]. We thought it useful to present the crystallographic structure analyses of (–)-**1a** and (+)-**1b**⁴⁾, in view of their very recently outlined structural differences with (–)-**2a,b** [9].



¹⁾ Derived from saccharine (= 1,2-benzisothiazol-3(2*H*)-one, 1,1-dioxide).

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³⁾ For recent reviews on asymmetric [4 + 2] and [3 + 2] intermolecular cycloadditions, see [3].

⁴⁾ For convenience, the arbitrary atom labels in **1a,b** are the same; for systematic names, see *Exper. Part*.

In both structures (–)-**1a** and (+)-**1b**, the *N*-enoyl side chain shows a typical *anti-s-cis* conformation ((–)-**1a**: S–N–C=O 148.8(4)°, O=C–C=C – 11.5(8)°; (+)-**1b**: S–N–C=O – 161.7(5)°, O=C–C=C – 13(1)°) as already observed for (–)-**2a,b**, although, for an identical absolute configuration, their bornane analogues have an opposite sign of the dihedral angle around the C(O)–C(α) bond ((–)-**2a,b**: O=C–C=C 1.0(9)° [5]; –6.1(8)° [6]). *Oppolzer* had suggested that the twisting of the dienophile around this bond might be important for the stereocontrol [8] [11], although no evidence of this influence could be found by recent PM3 calculations [9].

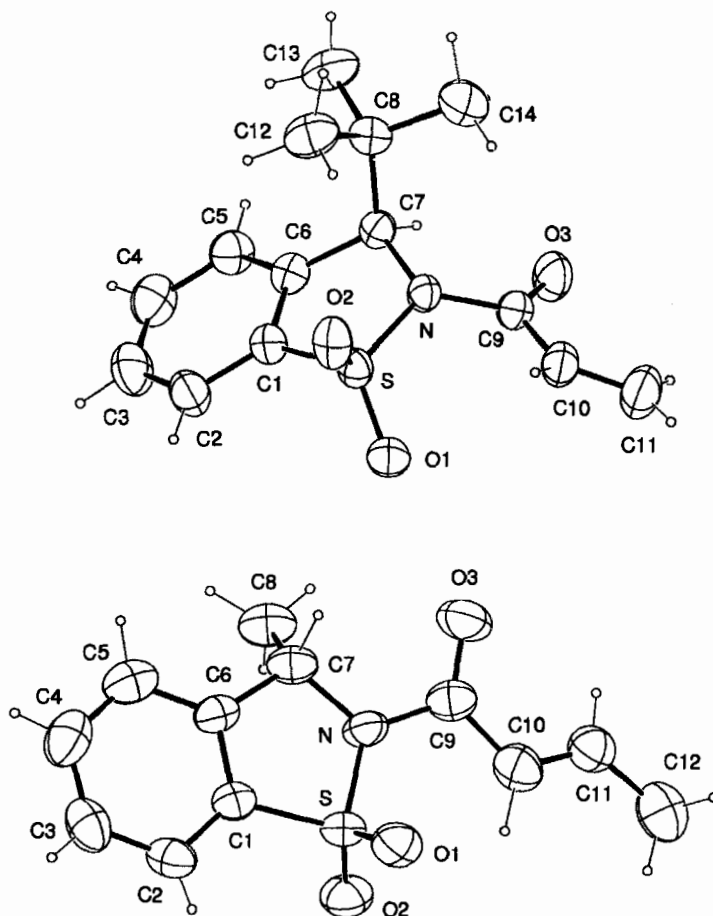


Figure. ORTEP Diagram of (–)-**1a** (top) and (+)-**1b** (bottom) with atom numbering (arbitrary). Ellipsoids are represented at the 40% probability level.

The *N*-acylbornanesultam derivatives (see **2**) systematically show a pseudoequatorial orientation for the C(2)–C(3) and S=O(2) bonds. For example, for (–)-**2a,b**, the S–N–C(2)–C(3) and O(2)–S–N–C(2) dihedral angles are 144.2(4) and 142.4(4)°, and –131.1(4)° and –125.9(4)°, respectively. This results from the *cis*-fused bicyclic five-

membered rigidified ring system as well as from the steric pressure of the Me(8) group on the O(2) in the bornanesultam skeleton. Indeed, in a pseudoaxial orientation, the O(2) atom would be much closer to Me(8) than the 3.446(9) and 3.230(9) Å observed in (–)-**2a,b**, respectively (for comparison, the distance between O(2) and C(12) in (–)-**1a** is 3.312(6) Å). This characteristic is not always respected in toluenesultams, since the flexible C(8) atom is not integrated in a ring, and because of the S-adjacent C(1) atom which is substituted and part of a benzene moiety. As a consequence, the aromatic ring exerts a *gauche* interaction on both pseudoequatorial alkyl and S=O(2) substituents and, to minimize the steric repulsion, tends to direct them in a pseudoaxial direction. For large alkyl groups, this results in the loss of Curran's postulated C_2 symmetry existing in (–)-**2a,b** [8]. This effect is particularly important for (–)-**1a**, where the *t*-Bu and S=O(2) substituents are frankly pseudoaxial (see Table 1), whilst the sterically less demanding Me substituent in (+)-**1b** is oriented in between the pseudoaxial (S–N–C(7)–C(8) *ca.* 100°) and the pseudoequatorial (*ca.* 130°) position. The aromatic ring is practically bisecting the O(1)=S=O(2) angle in the latter case (see Table 1).

Table 1. Selected Bond Lengths [Å] and Angles [°] for (–)-**1a** and (+)-**1b**

	(–)- 1a	(+)- 1b		(–)- 1a	(+)- 1b
C(1)–C(6)	1.376(6)	1.377(8)	O(2)–S–N	111.2(2)	111.3(3)
C(6)–C(7)	1.511(6)	1.489(9)	O(1)–S–C(1)	111.7(2)	112.0(3)
C(7)–C(8)	1.557(6)	1.51(1)	O(2)–S–C(1)	111.7(2)	112.1(3)
C(7)–N	1.496(5)	1.470(8)	N–S–C(1)	93.0(2)	93.5(3)
N–S	1.682(3)	1.671(5)	C(7)–N–C(9)	119.0(3)	120.5(5)
S–O(1)	1.432(3)	1.418(5)	C(9)–N–S	122.7(3)	123.2(5)
S–O(2)	1.425(3)	1.417(5)	C(7)–N–S	113.2(2)	114.6(4)
N–C(9)	1.414(6)	1.40(1)	O(3)–C(9)–N	119.2(4)	118.2(6)
C(9)–O(3)	1.201(6)	1.219(9)	O(3)–C(9)–C(10)	124.0(4)	124.0(7)
C(9)–C(10)	1.463(7)	1.45(1)	N–C(9)–C(10)	116.7(4)	117.7(6)
C(10)–C(11)	1.296(7)	1.30(1)	C(9)–C(10)–C(11)	120.7(5)	121.1(7)
C(1)–C(6)–C(7)	114.3(4)	115.5(6)	O(1)–S–N–C(7)	129.9(3)	–112.1(5)
C(6)–C(7)–N	104.6(3)	105.7(5)	O(2)–S–N–C(7)	–99.2(3)	118.2(5)
S–C(1)–C(6)	111.6(3)	110.6(5)	S–N–C(7)–C(8)	106.2(3)	–125.3(5)
O(1)–S–O(2)	116.6(2)	115.8(3)	C(2)–C(1)–S–O(1)	56.9(5)	–68.4(7)
O(1)–S–N	110.2(2)	110.0(3)	C(2)–C(1)–S–O(2)	–75.7(5)	63.6(7)

Another important feature of the cyclic sultams is the pyramidalization of the N-atom⁵⁾ and its indubitable correlation observed with respect to the S–N–C=O dihedral angle [9]. The N-lone pair (lp) is believed to be anomericly directed and stabilized, in the absence of major steric interactions, by the *anti*-periplanar S=O bond [9] [14], although, for *N*-acylbornanesultams, neither systematic lengthening of the S–N nor shortening of the S=O(1) bond lengths could be demonstrated [14]. According to this hypothesis, one would expect the N-lone pair to be *anti*-periplanar to the pseudoaxial S=O(2) bond in (–)-**1a**, but this would bring the *N*-enoyl chain close to both pseudoax-

⁵⁾ For planar and pyramidal acyclic sultam X-ray structure analyses, see [12] and [13], respectively.

ial substituents, thus resulting in a strong steric repulsion⁶). For this reason, the N-atom remains pyramidalized in the usual way ((-)-**1a**: $\Delta hN = 0.198(4)$ Å; (+)-**1b**: $\Delta hN = 0.112(7)$ Å), similarly to the *N*-enoylbornanesultam analogues (-)-**2a,b**.

In conclusion, the structural differences between *N*-enoyltoluene- and *N*-enoylbornanesultams are mainly due to the possible loss of the pseudo- C_2 symmetry. This results in a decrease of the stereoselectivity observed for the uncatalysed [4 + 2] cycloadditions of cyclopentadiene to (-)-**1a,b** at 21° (51% d.e. [1a]; 43% d.e., 75% *endo*, 57% yield) as compared to (-)-**2a,b** (66% d.e. [16]; 52% d.e. [16])⁷). In the case of bornanesultams (-)-**2a,b**, the steric approach is systematically directed onto the C(α)-*re* face for both *syn*- and *anti-s-cis* conformers, whilst this should be only the case for the highly reactive *syn-s-cis* (-)-**1a** conformer [14]⁸). Indeed, intuitively, according to Curran's postulate, a steric C(α)-*si* approach would be expected for (-)-**1a** in the *anti-s-cis* conformation⁹). Due to the small size of the Me substituent, resulting in a weaker *gauche* interaction with the aromatic ring and thus in a less pronounced C(8)/O(2) pseudodiaxial conformation, (-)-**1b** represents an intermediate case. Furthermore, as a result of the observed pyramidalization, the weak stereoelectronic preferred interaction¹⁰) is no longer mismatching the steric effect in the *anti-s-cis* (-)-**1a,b** conformers, in contrast to (-)-**2a,b** [9] [14] [19]. Supplementary X-ray structure analyses from cycloadducts derived from (-)-**1a,b**⁶), will be presented in due course.

Experimental Part

X-Ray Structure Determination of (-)-(3R)-3-(tert-Butyl)-2,3-dihydro-2-(1-oxoprop-2-enyl)-1,2-benzisothiazole 1,1-Dioxide ((-)-1a) and (+)-(3S)-2,3-Dihydro-3-methyl-2-[(E)-1-oxobut-2-enyl]-1,2-benzisothiazole 1,1-Dioxide ((+)-1b). Suitable crystals were grown from hexane/Et₂O and EtOH soln., resp. Cell dimensions and

- ⁶) For two X-ray structure analyses of cycloadducts with inverted N-pyramidalization, see [1b] for (3R)-*N*-[(1,2-dimethylcyclohex-3-en-1-yl)carbonyl]-3-methyltoluenesultam (= (3R)-2-[(1,2-dimethylcyclohex-3-en-1-yl)carbonyl]-2,3-dihydro-3-methyl-1,2-benzisothiazol 1,1-dioxide; $\Delta hN = -0.06(2)$ Å, O(2)–S–N–lp 177.6(9)°) and [15] for (3R)-*N*-[(1,2-dimethylcyclohex-4-en-1-yl)carbonyl]-3-methyltoluenesultam (= (3R)-2-[(1,2-dimethylcyclohex-4-en-1-yl)carbonyl]-2,3-dihydro-3-methyl-1,2-benzisothiazol 1,1-dioxide; $\Delta hN = -0.088(5)$ and $0.036(5)$ Å, O(1)–S–N–C(7) 136.0(3) and 139.0(3)°, O(2)–S–N–C(7) –91.6(3) and –89.0(3)°, S–N–C(7)–C(8) 96.6(4) and 99.3(4)°, and O(2)–S–N–lp 173.7(3) and 2.9(3)° for the two molecules of the asymmetric unit, resp.). In the latter case, the N-atom is practically planar with a lone pair almost symmetrically distributed on both π faces.
- ⁷) For a rationalization of the high diastereoselectivity observed in the presence of hypothetically unchelated dicoordinated (-)-**1a** during its [4 + 2] cycloaddition to cyclopentadiene, see [9].
- ⁸) For PM3 calculations showing the higher reactivity of the *syn-s-cis* conformer during the [3 + 2] cycloaddition of (-)-**1a**, see footnote 41 in [9]. This hypothesis better explains the observed stereoselectivity by competition with an *anti-s-cis* C(α)-*re* approach on the sterically more hindered face, shielded by the pseudoaxial *t*-Bu and O(2) substituents [17]. This is well highlighted by the [3 + 2] cycloadditions of acetonitrile and 2,2-dimethyl propionitrile oxides to (-)-**1a** (92% d.e.; 96% d.e., [2]) and (-)-**2a** (80% d.e.; 90% d.e., [5]), respectively.
- ⁹) For more precise PM3 calculations of [4 + 2] transition states of (-)-**1a**, see [9].
- ¹⁰) We believe that the stereoelectronic interaction is weaker than the steric interaction on the basis of the calculated transition-state energy for the cyclopentadiene cycloaddition to *N,N'*-fumaroylbis[(2*R*)-bornanesultam] [9] [18]. Indeed, comparison of the contra-steric bis(*syn-s-cis*) and bis(*anti-s-cis*) C(α)-*si* approaches, shows that, in contrast to (-)-**2a,b**, attack on the stereoelectronically favoured face, *anti* to the N-lone pair, is not always the origin of the minor diastereoisomer. Furthermore, the stereoelectronically non-additive *syn-s-cis-s-cis-anti* conformer has a lower activation energy than the bis(*syn-s-cis*) conformer for C(α)-*re* face attack.

Table 2. Summary of Crystal Data Intensity Measurement, and Structure Refinement for (–)-**1a** and (+)-**1b**

	(–)- 1a	(+)- 1b
Formula	C ₁₄ H ₁₇ NO ₃ S	C ₁₂ H ₁₃ NO ₃ S
Mol. wt.	279.4	251.3
Crystal system	orthorhombic	orthorhombic
Space group	<i>P</i> 2 ₁ 2 ₁ 2 ₁	<i>P</i> 2 ₁ 2 ₁ 2 ₁
<i>a</i> [Å]	10.5994(7)	7.274(2)
<i>b</i> [Å]	11.0282(10)	12.101(2)
<i>c</i> [Å]	12.2262(9)	13.943(4)
<i>V</i> [Å ³]	1429.2(2)	1227.3(5)
<i>Z</i>	4	4
<i>F</i> (000)	592	528
<i>D</i> _c [gr · cm ⁻³]	1.30	1.36
μ (MoK α) [mm ⁻¹]	0.219	0.247
((sin θ)/ λ) _{max} [Å ⁻¹]	0.64	0.58
No. measured reflections	1824	1147
No. observed reflections	1630	916
Criterion for observed	<i>F</i> _o > 2 σ (<i>F</i> _o)	<i>F</i> _o > 2 σ (<i>F</i> _o)
Refinement (on <i>F</i>)	full-matrix	full-matrix
No. parameters	172	154
Weighting scheme	$\omega = 1/(\sigma^2(F_o) + 0.0005(F_o^2))$	$\omega = 1/(\sigma^2(F_o) + 0.0006(F_o^2))$
Max. and min. $\Delta\rho$ [e · Å ⁻³]	0.27, –0.30	0.34, –0.38
<i>S</i>	1.34	1.23
<i>R</i> , ωR	0.054, 0.060	0.063, 0.054

intensities were measured at r.t. on *Philips-PW-1100* and *Nonius-CAD4* diffractometers with graphite-monochromated MoK α radiation ($\lambda = 0.71069$ Å). Data were corrected for *Lorentz* and polarization effects but not for absorption. The structures were solved by direct methods using MULTAN 87 [20], all other calculations used XTAL [21] system and ORTEP [22] programs. All H-atoms were observed and contributed to *F*_c calculations but were not refined. Table 2 shows details of the data collections and refinements. Crystallographic data have been deposited with the *Cambridge Crystallographic Data Center*, University Chemical Laboratory, 12 Union Road, Cambridge CB2 1EZ, England, as supplementary publication No. CCDC-10/54.

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